Effects of pH, particle size and porosity of raw rice husk and its silica on removing lead and hexavalent chromium from aqueous solution

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ABSTRACT
Adsorbent properties and aqueous characteristics are very important parameters in the removal efficiency (RE) of environmental pollutants. The main goal of this study was to investigate the effects of pH, particle size and porosity of raw rice husk and its silica on removing lead and hexavalent chromium (Cr\(^{+6}\)) from aqueous solution. The raw rice husk was collected from north of Iran and the rice husk silica was prepared at 800ºC after acid leaching. The effects of the adsorbent particle sizes (0.07-0.1, 0.1-0.5, 0.5-1.0 and 1.0-1.5 mm), porosity and pH from 2 to 8 were investigated by varying any of the process variables while keeping the other variables constant (adsorbent dosage = 1.5 g.\(^{-1}\), contact time = 60 min, chromium and lead initial concentration = 5 mg.\(^{-1}\)). The results of this study showed that the RE of Cr\(^{+6}\) is intensively pH - dependent. Using 1.5 g.\(^{-1}\) adsorbents, particle size = 0.5 - 1.0 mm, 5.0 mg.\(^{-1}\) initial concentration of Cr\(^{+6}\) and 60 min contact time, the maximum RE obtained by raw rice husk and its silica at pH 2 were up to 98.8 and 88.4%, respectively. However, at the same condition with changing pH (pH 7), the maximum RE decreased up to 69.4 and 60.4%, respectively. Moreover, a positive strong significant correlation was detected between decreasing the adsorbents particle sizes and lead removal efficiency in the two adsorbents (\(P < 0.01\)). The scanning electron microscope images of the two adsorbents showed that silica derived from raw rice husk has more porosity. In conclusion, the acidic condition of aqueous for Cr\(^{+6}\) and neutral condition for lead, increases the adsorbents porosity, while decreased adsorbent particle sizes causes an elevation in the RE of the two pollutants.

Key words: Hexavalent chromium, Lead; Particle size, Removal efficiency, Rice husk silica.

INTRODUCTION
Over the course of the recent decades, human activities have led to a significant increase in environmental pollution. This is also why control and treatment of pollutants are the most important environmental challenges today. Lots of studies have reported a relationship between environmental pollution levels and human health problems (Koop et al. 2010; Mehdinia 2011). Heavy metals are some of the most important environmental pollutants which have some hazardous effects on human health at exceeded concentrations (Kobya et al. 2004). Currently, several methods have been used to remove heavy metals from the environment. Some of the most important methods utilized in removing them work by adsorption (Abdel Ghani et al. 2007; Mehdinia et al. 2014). Activated carbon has been generally used for adsorption of heavy metals from aqueous. Since the use of commercial activated carbon to remove pollutants is expensive, investigators are searching for other cost-effective adsorbents. Recently, wide attention
has been devoted to the study of the heavy metals removal from water and wastewater with the application of agricultural wastes (Mahvi 2002; El-Ashtoukhy et al. 2004; Demirbas et al. 2004; Gueu et al. 2007; Shamohammadi et al. 2008; Montazeri et al. 2009; Khazaei et al. 2011; Salari et al. 2012; Kafia & surci 2012; Ghaneian et al. 2014; Mohseni et al. 2016). Adsorbent properties such as porosity, particle size, surface area and pore radius size as well as the properties of water and wastewaters such as pH and temperature are very important in determining the removal efficiency of pollutants. Therefore, this study investigates the effects of pH, particle size and porosity of raw rice husk and its silica on removing the lead and hexavalent chromium from water and wastewaters.

MATERIALS AND METHODS
Preparation of adsorbents
This research as an experimental study, was carried out in the School of Health, Semnan University of Medical Sciences in 2014. Raw rice husk was collected from the north of Iran. Rice husk silica was prepared according to Jamwal & Mantri method (2007). In this method, the rice husk was first washed with tap water to remove contaminants and was then oven-dried at 110°C for 24 h. It was then subjected to acid leaching by reflux in 3% hydrochloric acid and 10% sulphuric acid for 2 h, at a ratio of 50 g husk per liter. The husk was completely washed with distilled water and oven-dried at 100°C for 4 h. Eventually, the clean and dried husk was burned inside a muffle furnace at 800°C in static air for 4 h.

Batch adsorbent experimental
According to Souag et al. (2007), synthetic solution containing lead and chromium were prepared by dissolving a known amount of analytical-grade lead chloride and potassium dichromate in distilled water. The effects of the adsorbent particle sizes (0.07-0.1, 0.1-0.5, 0.5-1.0 and 1.0-1.5 mm), porosity and pH from 2 to 8 were investigated by varying any of the process variables while keeping the other variables constant (adsorbent dosage = 1.5 g.l⁻¹, contact time = 60 min, chromium and lead initial concentration = 5 mg.l⁻¹). The pH was adjusted from 2.0 through 8.0 by adding 0.1M HCl and 0.1M NaOH. All experiments were carried out in batch reactors at room temperature of 25 ± 2°C. All experiments were carried out at three times and the averages were reported as results.

Data analysis
The concentration of lead and hexavalent chromium before and after the adsorption process were measured using atomic adsorption (Perkin Elmer-100 model). Wavelengths of 283.3 and 357.9 nm and detection limits of 0.05 and 0.02 mg.l⁻¹ were used for measurement of lead and Cr⁶⁺, respectively. Removal efficiency of the adsorbents was determined by the following formula:

\[ RE = \frac{C_i - C_o}{C_i} \times 100 \]

where, \( C_i \) and \( C_o \) are the concentration of lead and hexavalent chromium before and after adsorption operating time, respectively (Mehdinia et al. 2011; Mehdinia et al. 2013). The surface morphology of the adsorbents were observed under a scanning electron microscope (SEM), using LEO operated at an accelerating voltage of 15 kV (Ros et al. 2007).

RESULTS AND DISCUSSION
Effect of particle size
The results of this study showed that the removal efficiency of hexavalent chromium is dependent on the particle sizes of raw rice husk and its silica. Using 1.5 g.l⁻¹ of the adsorbents, 5.0 mg.l⁻¹ initial concentration of Cr⁶⁺ and 60 min contact time, the maximum removal efficiency of up to 98.8 and 88.4% was obtained at particle size of 0.07-0.1 mm by the application of raw rice husk and its silica, respectively. However, at the same condition but with changing the particle size to 1.0-1.5 mm, the
maximum removal efficiency decreased up to 63.3 and 61.4%, respectively. The result of removal efficiency of Cr\textsuperscript{6+} using raw rice husk and its silica of different particle sizes and 5.0 mg.l\textsuperscript{-1} initial concentration, 60 min contact time and 1.5 g.l\textsuperscript{-1} adsorbents dosage are shown in Fig. 1. This result is supported by Choudhury \textit{et al.} (2012) who studied the adsorption of Cr (III) from aqueous solution using groundnut shell. They reported that: 'The lower the particle size, the higher the chromium removal (%). By decreasing in particle size, the removal of chromium (III) increased from 47 to 97%'. Moreover, Wong \textit{et al.} (2003) and Munaf & Zein (1997) reported that by elevating the adsorbent particle size, the removal efficiency of heavy metals will be decreased.

![Fig. 1. Removal efficiency of Cr\textsuperscript{6+} using raw rice husk and its silica with different particle sizes, 5.0 mg.l\textsuperscript{-1} concentration, 60 min contact time and 1.5 g.l\textsuperscript{-1} adsorbent dosage.](image)

In addition, in this research, we studied the correlation between decreasing adsorbents particle size and removal efficiency of lead using the Pearson’s correlation analysis. So that, we assessed the correlation between decreasing the raw rice husk and its silica particle sizes (0.07-0.1, 0.1-0.5, 0.5-1.0 and 1.0-1.5 mm) as well as removal efficiency of lead at pH 6, adsorbents dosage = 1.5 g.l\textsuperscript{-1}, contact time = 60 min and initial concentration of lead = 5.0 mg.l\textsuperscript{-1}. A positive strong significant correlation was detected between decreasing the adsorbent particle sizes and lead removal efficiency in both adsorbents (P < 0.01). The results showing the correlation between decreasing particle sizes and lead removal efficiency is illustrated in Fig. 2. These effects could be due to the fact that the smaller adsorbent particles offer larger surface areas and greater numbers of adsorption sites (Choudhury \textit{et al.} 2007).

**Effect of pH**

The results of this study showed that the removal efficiency of hexavalent chromium is intensively pH-dependent. Using 1.5 g.l\textsuperscript{-1} adsorbents, 5.0 mg.l\textsuperscript{-1} initial concentration of Cr\textsuperscript{6+} and 60 min contact time, the maximum removal efficiency of 98.8 and 88.4% by the application of raw rice husk and its silica, were obtained at pH 2, respectively. However, at the same condition only by changing pH values, the removal efficiency decreased as the maximum RE at pH 7 reached up to 69.4 and 60.4, respectively. Unlike chromium, lead removal efficiency is not so pH-dependent. Using 1.5 g.l\textsuperscript{-1} of the adsorbents, particle size = 0.5-1.0 mm, 5.0 mg.l\textsuperscript{-1} initial concentration of lead and 60 min contact time, maximum removal efficiency of 69 and 97.4% were obtained by the application of raw rice husk and its silica, respectively at pH 6. Therefore the removal efficiency did not
Change by changing pH values. Hasan et al. (2004) studied the removal of hexavalent chromium from aqueous solutions using agricultural waste ‘maize bran’. They reported that the initial removal of hexavalent chromium increased by an increase in pH from 1.4 to 2.0, thereafter removal started to decrease by elevating pH from 2.0 to 8.5. This could be that since hexavalent chromium occurs in the form of oxyanions for example HCrO$_4^-$, Cr$_2$O$_7^{2-}$, CrO$_4^{2-}$, etc. in acidic medium and the lowering of pH causes the surface of the adsorbent to be protonated to a higher amount, consequently a strong attraction exists between these oxyanions of hexavalent chromium and positively-charged surface of the adsorbent. Hence, the removal efficiency increases by a decrease in the pH of the aqueous (Hasan et al. 2008).

Demirbas et al. (2004) studied the removal of hexavalent chromium by three types of carbon at different pH levels at an initial concentration of 105 mg.l$^{-1}$ and particle sizes of 1.0 to 1.25 mm. They reported that optimum removal efficiency was observed at pH 1.0. They also reported that the dominant form of hexavalent chromium at pH 2 is HCrO$_4^-$. Increasing in the pH will shift the concentration of HCrO$_4^-$ to other forms, CrO$_4^{2-}$ and Cr$_2$O$_7^{2-}$.

The results of lead and hexavalent chromium removal efficiency using raw rice husk and its silica with different solution pH, 5.0 mg.l$^{-1}$ initial concentration, 60 min contact time and 1.5 g.l$^{-1}$ adsorbent dosages are shown in Fig. 3.

![Fig. 2. The correlation between decreasing in particle sizes and lead removal efficiency at pH 6, adsorbents dosage of 1.5 g.l$^{-1}$, initial concentration of lead = 5 mg.l$^{-1}$ and contact time of 60 min by the two adsorbents.](image)

**Effect of adsorbent morphology**

Scanning electron microscope (SEM) shows the morphologic images of the raw rice husk and the silica derived from modified rice husk. It showed that when the rice husk goes under on acid leaching and incineration at high temperature, it reduces the crystallization of cells and increases its porosity. This phenomenon can thus increase the potential of adsorbents to remove environmental pollutants. Mehrasbi & Farahmandkia (2008) reported that modification of banana shell using acidic an alkaline agents results in increased porosity and surface area of the adsorbent (Mehrasbi & Farahmandkia, 2008). This result is also supported by Wang Ngah & Hanafiah (2008) who investigated the removal of the heavy metal ions from wastewater using chemically-modified plant wastes as adsorbents.

In order to compare the shape and pores of the raw rice husk with its silica, SEM images of rice
husk silica with magnification of (a) 1000 and (b) 2000 are shown in Fig. 4 while the SEM images of raw rice husk with magnification of (c) 500 and (d) 1000 are shown in Fig. 5.

**Fig. 3.** Removal efficiency of lead and Cr\(^{6+}\) using raw rice husk and rice husk silica with different solution pH and 5.0 mg.l\(^{-1}\) initial concentration, 60 min contact time and 1.5 g.l\(^{-1}\) adsorbents dosage.

**Fig. 4.** Scanning electron microscope (SEM) images of silica derived from raw rice husk with magnification of 1000 (a) and 2000 (b).

**Fig. 5.** Scanning electron microscope (SEM) images of raw rice husk with magnification of 500 (c) and 1000 (d).
CONCLUSION
The results of this study showed that the removal efficiency of hexavalent chromium and lead from aqueous with the application of raw rice husk and its silica increase using smaller particle sized adsorbents. A strong positive significant correlation was observed between decreasing adsorbents particle sizes and lead removal efficiency. According to the results of this study, acidic condition of aqueous for Cr(Ⅵ) and neutral condition for lead, by increasing in adsorbents porosity and decreasing in adsorbents particle sizes result in elevation of the removal efficiency by the both adsorbents.

Competing interests
The authors declare that they have no competing interests.

ACKNOWLEDGEMENTS
This article derived from research project no. 418 in Semnan University of Medical Sciences and Research University Grant (RUGs) through Project No: 91806 in University Putra Malaysia (UPM). Therefore, authors would like to acknowledge Semnan University of Medical Sciences and University Putra Malaysia for the financial supports.

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بررسی تأثیر pH اندازه ذرات و تخلخل شلتوک خام برنج و سیلیکای حاصل از آن در حذف کروم شش ظرفیتی و سرب از محیط‌های آبی

معینیان خ. 1، عبداللطیف ف.ب. 2، راستگو ط. 1، مهدی نیا س.م. 1

چکیده
خصوصیات مواد جاذب و ویژگی‌های محلول پارامترهای بسیار مهمی در کارایی حذف آلاینده‌های زیست محیطی می‌باشند.
هدف این مطالعه بررسی تأثیر pH اندازه ذرات و تخلخل شلتوک خام و سیلیکای حاصل از آن در حذف سرب و کروم شش ظرفیتی از محیط‌های آبی بوده است. شلتوک خام از منطقه شمال ایران جمع‌آوری گردید و سیلیکای شلتوک بجع پس از مرحله اسید شویی و احترق در دمای 033 (درجه سانتی گراد) در کوره تهیه گردید. تاثیر اندازه ذرات جاذب (مرحله 100-200 میلی متر)، تخلخل و pH محلول از 2 تا 8 مورد مطالعه قرار گرفت. در این بحث تحقیق هریک از متغیرها با ثابت نگه داشتن سایر پارامترها (جرم مواد جاذب 9/1 گرم در لیتر، مدت زمان تماس 93 دقیقه و غلظت اولیه سرب و کروم برابر با 9/3 میلی گرم در لیتر) مورد بررسی قرار گرفتند. نتایج این تحقیق نشان داده است که راندمان حذف کروم شش ظرفیتی به شدت pH وابسته می‌باشد. با کاربرد 15 گرم در لیتر مواد جاذب، با ابتدای ذرات 15/000 گرم pH وابسته به شدت pH در زمان ماندگاری 9/1 دقیقه، راندمان حذف برای شلتوک خام و سیلیکای حاصل از آن به ترتیب برای pH 8/8 و 9/4 درصد در pH حاصل از آن برابر با pH 9/4 و 8/3 درصد در pH حاصل گردیده‌است که کاهش حاصل از آن به کاهش حاصل از آن برابر با بهبود و قوی بین کاهش اندازه ذرات مواد جاذب و افزایش راندمان حذف سرب و وجود دارد (P < 0/01). بررسی میکروسکوپ الکترونی مواد جاذب نشان داده است که سیلیکای حاصل از شلتوک بجع دارای میزان غلظت 9/1 برابر به دسته خام می‌باشد. مطالعه حاضر نشان داده است که شرایط اسیدی محیط‌های آبی برای حذف کروم شش ظرفیتی و شرایط خشک برای حذف سرب، افزایش میزان تخلخل مواد جاذب و کاهش اندازه ذرات مواد جاذب باعث افزایش راندمان حذف این آلاینده می‌گردد.

(تاریخ دریافت: 03/01/1395)